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# **Electron-Spin Resonance (ESR)**

## Abstract

Electron-spin resonance (ESR) spectroscopy deals with the *interaction of electromagnetic radiation with the magnetic moment of unpaired electron spins*. Due to its versatility ESR has found applications in a wide range of fields, from chemistry and biology to the novel and stimulating areas of quantum computation and "single-spin" detection.

#### **Objectives:**

- · Learn the basics of microwave spectroscopy.
- Learn the use and applications of lock-in amplifiers.
- By carrying out simple ESR measurements, determine the electron *g*-factor.





E. K. Zavoisky (1907-76) discoverer of the ESR

**Fig. 1:** (a) The electron-spin Zeeman effect: progressive splitting of the  $M_s = \pm \frac{1}{2}$  spin states in an applied magnetic field. Only when the field is swept across the resonance ( $\mathbf{D}E = g\mathbf{m}_{_B}H_0 = h\mathbf{w}$ ), there is a transfer of energy from the EM radiation (here at constant  $\mathbf{w}$ ), giving rise to an absorption line (b). In ESR the resonance is observed by detecting its derivative  $d\mathbf{c}''/dB$  (c).

## **Experimental approach**

To detect the electron-spin resonance a DPPH sample (Fig. 3, inset) is positioned in a microwave cavity. Two sets of coils (Fig. 2) generate a (large) *static*  $H_0$  and a (small) *modulating*  $H_m$  magnetic field, respectively. When the main field is swept across the resonance, the sample absorbs EM radiation, hence spoiling the balance of the microwave bridge, previously tuned to provide a nil response in the microwave detector. At resonance, the weak signal generated in the diode detector is captured by the lock-in amplifier, whose output is recorded and displayed in a computer. Successive analyses provide the *width* and *position* of the ESR resonance, from which the g-value of the electron (typical of a given material) can be estimated.



**Fig. 2:** Simplified diagram of the ESR spectrometer. The waveguide system includes also a phase shifter, a hybrid junction (shown in detail) and two attenuators. Arrows indicate the propagation direction of the microwaves.



**Fig. 3:** Structure of the stable free-radical DPPH (2,2-diphenyl-1picrylhydrazyl) (inset) and its ESR spectrum, as measured in the current experiment. A Gunn diode was used as a source of the Xband (9.75 GHz) radiation.

### Results

The X-band ESR spectrum of a DPPH sample, measured in normal conditions (after magnet calibration), is shown in Fig. 3. The slightly asymmetric line-shape is due to the large sample volume. The evaluated *g*value, 2.012(2), is within 0.4% from the accepted reference, and similar to that of free electrons.

#### **Further Reading**

- C. P. Poole Jr., *Electron Spin Resonance* (Wiley, New York, 1983).
- D. M. Pozar, *Microwave Engineering*, 2<sup>nd</sup> ed. (Wiley, NY, 2001).
- J. A. Weil and J. R. Bolton, *Electron Paramagnetic Resonance*, 2<sup>nd</sup> ed. (John Wiley & Sons, Hoboken, NJ, 2007).
- M. R. Brustolon and E. Giamello, eds. *Electron Paramagnetic* Resonance: A Practitioners Toolkit (Wiley Hoboken, NJ, 2009).